Abstract
Methemoglobinemia is an alteration of the oxidative state of hemoglobin. When methemoglobin values rise above 10%, the symptoms and signs related to this condition appear, such as cyanosis, respiratory problems, fatigue and headache. Acquired methemoglobinemia is characterized by the sudden onset of cyanosis in a previous healthy child, which can be due to exposure to medications or chemical substances including nitrates, copper, sulfates, chlorites, chloramines and chlorates which can be present in food and water. We illustrate two cases of acquired methemoglobinemia related to nitrate ingestion from a vegetable source.

Introduction
Methemoglobinemia is a rare blood cells disorder due to the hemoglobin iron oxidation from a ferrous to a ferric state. Methemoglobin (MET) cannot carry and release oxygen, with cyanosis and tissue hypoxemia as the consequence. Normal MET concentration in the blood is less than 2%.

Methemoglobinemia can be congenital or acquired. Congenital methemoglobinemia is characterized by early onset of cyanosis without any cardiac and pulmonary disease. Acquired methemoglobinemia is characterized by sudden cyanosis onset in a previous healthy child and can be due to exposure to medications, food and water. Acute diarrhea may also cause oxidation of hemoglobin to methemoglobin (Table 1).

Although Acquired methemoglobinemia has been well described before, this represents a challenging condition for any clinicians. Indeed, in order to avoid long-term consequences and reduce the risk of repeated episodes, it is very important to identify this disease and carry out an accurate history to identify its etiology.

We present two cases of acquired methemoglobinemia related to nitrate ingestion from a vegetable source, in order to remind practitioners that the acute onset of cyanosis can depend on other factors than cardiovascular and/or respiratory disease.

Case Reports
Case #1
A 10-years-old boy was admitted to our emergency department for perioral cyanosis and bluish nail bed discoloration, associated with abdominal pain and tenderness. His mother reported an episode of lipothymia 30 minutes before. There was no history of significant family medical conditions or previous illnesses.

On admission, the patient was in good general conditions: blood pressure was 90/70 mmHg, heart rate 100/min, respiratory rate 35/min, oxygen saturation in room air 92%, capillary refilling time was 2” and body temperature 36°C.

Physical examination showed a perioral cyanosis and bluish discoloration of the nail bed. Cardio-thoracic and abdominal examinations were unremarkable Electrocardiogram and chest X-ray showed no abnormalities, as well as there were no pathologic findings at echocardiography.

Blood tests revealed white blood cells 15,720/mL with normal differential count, platelet (PLT) 316,000/mL, hemoglobin (Hb) 13.9 g/dL, Hematocrit (Ht) 38.9%, erythrocyte sedimentation rate (ESR) 15 mm/1h (normal value 20 mm/1h), C-reactive protein (CRP) 1.3 mg/L; The value of MET using multiple-wavelength co-oximetry was reported to be 15%.

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CO-oximetry was 15%, confirmed by the venous blood gas (MET=15%). During a further assessment of recent drug and diet history, the mother reported a significant assumption of spinach during the last meal. The diagnosis of diet-induced methemoglobinemia was made and oxygen administration via nasal cannula was sufficient to treat this condition. After 2 hours of treatment, hemogasanalysis revealed significantly reduced MET levels (1.5%). The patient was discharged on the following-day without prescription of any home-medication. He returned for a check-up a month later and was in good health.

Case #2

An 8-months-old child was admitted to our emergency department with cyanosis and vomiting, suddenly onset about 40 minutes after his last meal. The child had no significant family medical conditions or past illnesses. His parents revealed that their son was not on any medication.

On admission, the patient was very irritable, with blood pressure at 85/60 mmHg, heart rate 150/min, respiratory rate 45/min, and oxygen saturation in room air 90%, while capillary refilling time was 2’ and body temperature 36.3°C. Physical examination showed perioral and finger area cyanosis. Cardiothoracic and abdominal examinations were unremarkable.

Most common causes of central and peripheral cyanosis were ruled out by echocardiography and electrocardiogram. Chest-X ray was normal. Blood samples were noticed to have a chocolate color (Figure 1), rising the suspect of acquired methemoglobinemia. On further interrogation, the mother revealed that the child has assumed a vegetable soup, mainly composed of tomatoes, spinach and potatoes which had been prepared two days before and kept in a refrigerator.

As per our hypothesis, multiple-wavelength CO-oximetry showed high MET levels (21%), confirmed by the venous blood gas (22%). All other blood tests were negative. The diagnosis of methemoglobinemia was made and the patient was treated with intravenous injection of methylene blue (1 mg/kg). After 1 hour, hemogasanalysis showed reduced MET (0.8%), while oxygen saturation in room air was a steady 99%. The patient was discharged two days later without any medications. After one month, he returned for the follow-up and was found to be in good health.

Discussion

Acquired methemoglobinemia is more frequently encountered than the congenital form and is frequently due to the exposure to hemoglobin oxidants, which overwhelm the cytochrome-b5 reductase’s ability to reduce ferric hemoglobin. As a consequence, hemoglobin iron is transformed into ferric ion (Fe³⁺) that is unable to transport oxygen. Common agents causing acquired methemoglobinemia include direct oxidizing agents (e.g. benzocaine and prilocaine), metabolic activation (e.g. aniline and dapsone) or indirect oxidation (e.g. nitrates).

Ash-Bernal et al. described eleven pediatric patients among a retrospective series of 138 cases with acquired methemoglobinemia registered in two different hospitals. They showed that the most common causes of methemoglobinemia in their cohort were exposure to drugs and gastrointestinal infections with dehydration. Almost all (94%) patients with methemoglobinemia were anemic.

Drugs are widely described as a cause of methemoglobinemia, with the most common including dapsone and topical anesthetic agents.

Table 1. Hereditary and acquired methemoglobinemia.

<table>
<thead>
<tr>
<th>Hereditary methemoglobinemia</th>
<th>Acquired methemoglobinemia</th>
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<tbody>
<tr>
<td>Cytochrome b5-reductase deficiency</td>
<td>Local anesthetics: benzocaine, lidocaine</td>
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<tr>
<td>Hemoglobin M</td>
<td>Antimicrobials: chloroquin, dapsone, primaquin, sulfonamides</td>
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<tr>
<td>Analgesics: phenacetin, phenazopyridine</td>
<td>Nitrites: contaminated well water or from diet. Fires and smoke inhalation.</td>
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<tr>
<td>Other: aminophenol, aniline, broomates, chloramphenicol, nitrobenzene, chloride benzene, nitroglycerin, nitrous oxides</td>
<td>Diarrhea</td>
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<tr>
<td>Acidosis</td>
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Figure 1. Chocolate color blood.
Dapsone is a drug used both as prophylactic therapy against *Pneumocystis Carinii* in severely immunocompromised patients and as therapy for leprosy and dermatitis herpetiformis. Dapsone-induced methemoglobinemia may be dose-related or an idiosyncratic reaction in susceptible individuals. Chegondi *et al.* described a case of a 4-years-old girl with end stage renal failure who developed a methaemoglobinemia during therapy with Dapsone given as prophylactic therapy against *Pneumocystis Carinii.* After having been absorbed in the gut, this drug is metabolized into several metabolites, including dapsone hydroxylamine, an oxidant agent that can be responsible for methemoglobinemia and hemolytic anemia. In patients with renal failure, the impaired excretion of dapsone metabolites makes them more susceptible to methemoglobinemia.

A recent review from Guay *et al.* summarized 242 individual episodes of methemoglobinemia due to prilocaine or benzocaine use in 233 patients (9 repeated episodes) published between 1949 and 2007. These patients included 80 children (44 infants 6-months-old or less). Forty patients had an intoxication related to Benzocaine and 37 to Prilocaine.

In 2016, Cho *et al.* described a case of a 3.5-years-old child with seizures and methemoglobinemia after topical application of EMLA, an eutectic mixture of lidocaine and prilocaine widely used topically to provide local anesthesia for a variety of painful superficial procedures.

Ingestion of nitrates is a rare cause of acquired methemoglobinemia in children. Usually, it is due to the ingestion of contaminated water or vegetables such as beets, spinach, peas and carrots. Variations in the chemical composition of plants can be linked both to season, soil composition, farming, and type and quantity of fertilizers used, or to inadequate transport or packaging conditions.

Sanchez-Echaniz *et al.* presented the clinical and epidemiological data regarding 7 infants with acquired methemoglobinemia related to consumption of vegetables. They mean age of patients was 8.14 months and none of the infants had diarrhea or was undernourished. In all cases, there was an history of assumption of homemade puree of mixed vegetables prepared in advance and kept in the refrigerator for 12 to 27 hours. Silver beets were a common ingredient and the methemoglobin levels ranged between 10% and 55%. Three infants had nitrituria, which suggested the diagnosis of methemoglobinemia.

Children at higher risk of intoxication are those below the age of 6 months, due to the reduced gastric acid production, a gut microbiota rich in nitrate-reducing bacteria, an immature methemoglobin reductase system, and the presence of certain amount of fetal hemoglobin that is more easily oxidized.

Whereas in the second case described, both the age and the food history of the child quickly suggested the diagnosis, the first case that we described concerned a 10-years-old boy, which reminds us that, although unusual, this condition can occur at any age, especially in situations in which there is less availability of fresh food and therefore people consume more processed food.

It is actually well known that sodium nitrate and sodium nitrite are used for their antimicrobial effects to preserve and cure meat, fish, and certain cheeses. Few cases of methemoglobinemia due to accidental ingestion of high concentration of preservatives are described in literature. Kennedy *et al.* described a family outbreak of methemoglobinemia following ingestion of sausages made using ‘saltpetre’ a compound based on potassium and sodium used in cooking.

It has to be taken in count that also the nasal inhalation of volatile nitrates can cause methemoglobinemia.

The volatile nitrates are sweet-smelling liquids normally administered by nasal inhalation and commonly present in some illicit recreational drugs known as poppers. The nitrite present in poppers is amyl nitrite, an agent capable to dilate blood vessels induing tachycardia and an euphoric feeling. In a study from 2000 and 2001 by National Household Surveys on Drug Abuse, 1.5% of interviewed adolescents aged 12 to 17 years reported any lifetime use of nitrite inhalants.

So, the intentional assumption of drugs should be considered when methemoglobinemia is suspected.

Physiologically, levels of methemoglobin are equal to 2% in term babies and less than 2% in adults. Cyanosis appears when methemoglobinemia levels are between 10% and 20%. Values between 20% and 50% are associated with respiratory problems, dizziness, headache, and fatigue. Values above 60% can be lethal.

On admission, our patients had methemoglobin levels of 15% and 22% respectively, presenting with mild symptoms.

If methemoglobin serum levels cannot be measured, the presence of an oxygen ‘saturation gap’ between pulse oximetry and arterial blood gas analysis measurements is suggestive of methemoglobinemia. Indeed, pulse oximetry exclusively detects oxyhemoglobin and deoxyhemoglobin, while arterial blood gas analysis detects also methemoglobin, thus providing higher saturation readings. When the observed gap is superior to 5 percentage saturation points, the presence of forms others than oxy- and deoxyhemoglobins should be suspected (i.e. methemoglobin, carboxyhemoglobin).

Moreover, in presence of low saturation registered at standard pulse oximetry, an arterial blood gas sample should be always performed.

However, in 2005, Masimo Corp. developed the new ‘Rainbow Technology’ pulse oximeters permitting the non-invasive measurement of carboxyhemoglobin, methemoglobin, and total hemoglobin as this device uses eight wavelengths of light corresponding to specific characteristic of absorbance values of derivatives of hemoglobin. The additional wavelengths improve the accuracy of the spectrum, minimizing or eliminating the interfering substances and better detect the presence of the desmoglobin. So, when available, multiple-wavelength co-oximetry is the best choice to diagnose methemoglobinemia.

Forms with methemoglobin <20% can be treated by oxygen administration and elimination of the source of contamination from the diet. In symptomatic patients with methemoglobin levels higher than 20%, or in presence of MET levels higher than 30%, treatment with methylene blue administration is suggested.

Methylene blue, slowly injected intravenously at the dose of 1-2 mg/Kg, combines with nicotinamide adenine dinucleotidedephasphatereduced (NADPH), in the presence of NADPH-methemoglobinreductase, to produce leukomethylene blue which then reduces the heme group acting as cofactor to accelerate the conversion of methemoglobin to hemoglobin in erythrocytes. Methylene blue can reduce the half life of methemoglobin from hours to minutes. Because glucose 6 phosphate dehydrogenase (G6PD) is required to regenerate NADPH, Methylene blue therapy is ineffective in individuals with G6PD deficiency. Moreover, in these patients Methylene Blue can cause severe hemolysis itself an oxidant agent.

**Conclusions**

In conclusion, acquired methemoglobinemia should be consid-
ered in case of acute cyanosis in children with no respiratory symptoms or cardiovascular disease. A good history and high level of suspicion are required to make the diagnosis and the use of multiple-wavelength co-oximetry can be a non-invasive instruments for the diagnosis of methemoglobinemia.

References