SHORT COMMUNICATION

Viburnum opulus: Could it be a new alternative, such as lemon juice, to pharmacological therapy in hypocitraturic stone patients?

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Summary

Objective: Citrate, potassium, and calcium levels in Viburnum opulus (V. opulus) and lemon juice were compared to evaluate the usability of V. opulus in mild to moderate level hypocitraturic stone disease.

Materials and Methods: V. opulus and lemon fruits were squeezed in a blender and 10 samples of each of 100 ml were prepared. Citrate, calcium, sodium, potassium, magnesium, and pH levels in these samples were examined.

Results: Potassium was found to be statistically significantly higher in V. opulus than in lemon juice (p = 0.006) whereas sodium (p = 0.004) and calcium (p = 0.008) were found to be lower. There was no difference between them in terms of the amount of magnesium and citrate.

Conclusions: Because V. opulus contains citrate as high as lemon juice does and it is a potassium-rich and calcium- and sodium-poor fluid, it can be an alternative to pharmacological treatment in mild-to-moderate degree hypocitraturic stone patients. These findings should be supported with clinical studies.

KEY WORDS: Viburnum opulus; Hypocitraturic; Urinary stone; Lemon juice.

Submitted 24 October 2014; Accepted 31 October 2014

INTRODUCTION

Urinary tract stone disease affects 12% of the world’s population and its recurrence can be as high as 50% at 10 year follow up (1). With the widespread use of SWL and introduction of endoscopic techniques, studies on medical stone treatment have been reduced and pushed to the background (2).

Urinary tract stones are formed through a marked increase in the saturation of a solute substance in the urine. One of the changeable factors affecting solubility is pH. Increase in the value of urinary pH raise the point of solubility and may prevent stone formation (3). The agents commonly used to treat hypocitraturia and alkalizing urine are sodium citrate and potassium citrate (4, 5). Many studies have shown that citrate replacement reduces rates of stone recurrence. However, patient compliance to pharmacotherapy with potassium citrate can be difficult. Due to gastrointestinal side effects and the high number of tablets to take throughout the day, 3-year treatment dropout rates of patients are as high as 25 percent (2, 5). Natural citric acid intake can be used as an alternative to pharmacotherapy in patients who are incompatible, or who cannot tolerate potassium citrate. There are studies that have used lemonade, orange, grapefruit, lime, or tomato to aim this target (6-11). These studies emphasize that they can be a good alternative to pharmacotherapy in mild to moderate hypocitraturia. Oxidative stress and renal tubular cell injury are observed in urinary tract stone patients. Lipid peroxidation begins in the cell membrane as a consequence of the toxicity of free radicals. When cell membrane integrity breaks down, cell balance begins to disappear, and cell death starts. Antioxidants may be used in order to avoid this situation (12). Viburnum opulus (V. opulus) has antioxidant properties (13-15) and we think that it may have a place in the medical treatment of stone disease. Not only with its antioxidant properties, but also with its content of potassium and citrate, it suggests being beneficial in the prevention of stone disease. In our study, V. opulus is compared in terms of citrate, calcium, phosphorus, magnesium, sodium, potassium and pH to lemon, which is known as a source of natural citrate, and the usability of V. opulus in hypocitraturic stone disease patients is discussed.

MATERIAL and METHOD

Because V. opulus has a bitter taste when first collected, the fruit was kept in brine water for a month to make it lose its bitterness. After having made it drinkable, it was squeezed in a blender and 10 samples of each of 100 ml were collected. Ten samples of 100 ml each were taken from lemons in the same manner. During evaluation, no dilutions were made with water or any other solution and no sugar was added to prevent the direct effect of a liquid or sugar to the variables to evaluate. The resulting extract was centrifuged for 15 min at 2000 g. Citrate, oxalate, calcium, phosphorus, magnesium, sodi-
um, potassium, chloride, and pH levels were examined in all samples.
Samples were grouped as follows: Group 1. *V. opulus* fruit held in brine; Group 2. Fresh lemon juice.
An enzyme-spectrophotometric method was used to determine citrate in biological fluids. It is based on citrate lyase and phenylhydrazine reactions. The enzyme converts citrate into oxaloacetate, which, in the presence of phenylhydrazine, is transformed into the corresponding phenylhydrazone. The ultraviolet-absorbing product is determined by absorbance measurement at 330 nm (16). The method is based on the following reaction:

\[
\text{Citrate lyase} \quad \text{Citrate} \rightarrow \text{oxaloacetate + acetate}
\]

\[
\text{Malate dehydrogenase} \quad \text{Oxaloacetate + NADH + H}^+ \rightarrow \text{malate + NAD}^+
\]

Calcium, phosphorus and magnesium (Roche Diagnostics GmbH, Mannheim) were colorimetrically measured by a Hitachi P800 autoanalyser (Hitachi High-Technologies Co., Japan). Sodium and potassium were analyzed in the Hitachi P800 autoanalyser using ion-selecting electrodes. A pH meter Precisa pH 900 device was used to verify pH. Wilcoxon Rank-Sum test was used in the statistical evaluations for comparison of the parameters.

**Results**

The citrate and magnesium contents of *V. opulus* have been found not statistically different from than of lemon juice. Potassium has been found statistically higher than in lemon juice (p: 0.006) whereas sodium (p: 0.004) and calcium (p: 0.008) were lower. Table 1 shows the citrate, calcium, sodium, potassium, phosphorus, magnesium and pH values in lemon juice and *V. opulus* contents.

**Table 1.**

<table>
<thead>
<tr>
<th>Solute</th>
<th>Lemon juice</th>
<th>V. opulus</th>
</tr>
</thead>
<tbody>
<tr>
<td>Citrate (mmol/L)</td>
<td>65.22 ± 5.86</td>
<td>54.04 ± 5.05</td>
</tr>
<tr>
<td>Potassium (mmol/L)</td>
<td>40.51 ± 2.78</td>
<td>27.55 ± 2.12</td>
</tr>
<tr>
<td>Calcium (mmol/L)</td>
<td>0.05 ± 0.01</td>
<td>1.52 ± 0.02</td>
</tr>
<tr>
<td>Magnesium (mmol/L)</td>
<td>1.57 ± 0.26</td>
<td>1.44 ± 0.21</td>
</tr>
<tr>
<td>Sodium (mmol/L)</td>
<td>2.54 ± 0.19</td>
<td>6.35 ± 0.98</td>
</tr>
<tr>
<td>pH</td>
<td>4.02 ± 0.16</td>
<td>4.03 ± 0.18</td>
</tr>
</tbody>
</table>

**Discussion**

A well known inhibitor of calcium-based stones is citrate. Citrate reduces calcium oxalate and phosphate saturation by forming calcium-soluble complexes and by inhibiting crystal nucleation and growth (4). With an incidence of 16-63%, hypocitraturia is an important etiological factor in recurrent calcium nephrolithiasis (3). It has been demonstrated that pharmacological potassium citrate intake increases urine citrate levels and reduces urine calcium excretion as well as relative saturation of calcium oxalate in hypocitraturia (2). Alternatives to potassium citrate have been sought in recent years due to patient non-compliance, particularly due to gastrointestinal poor tolerance (17-45%), and to the severe financial burden (a daily dose price of $ 3.90 in Turkey) (5). Significant increases in urinary volume, pH, potassium, magnesium and citrate excretion have been obtained in patients with hypocitraturic stone disease by the addition of fruit and vegetable juices to the diet (7, 17, 18). In addition, this has also been reported to provide dilution of lithogenic risk factors in the urine without affecting the concentration of potassium and citrate (19). In particular, various studies have established that the use of citrate extracts and juices as a natural source of citrate can be an alternative to potassium citrate (6, 8). It has been shown that high concentrations of citrate in citrus products may affect urine citrate excretion (20, 21). Orange juice causes an alkali load by increasing net gastrointestinal alkali absorption, increases urinary pH and citrate and reduces ammonium and net acid excretion. It is also reported that daily consumption of one liter of orange juice increases citraturia and pH and prevents stone formation and reduces crystallization risk factors for calcium phosphate (8, 22, 23).

Although grapefruit juice has been shown to have higher citrate content, it has not been possible to demonstrate that it reduces urinary risk factors (24, 25). In addition, grapefruit juice may affect metabolism of commonly used drugs by inhibiting the cytochrome P-450 (24). The citraturic effect of lemonade has been established by a variety of studies. An advantage of lemonade to orange juice is that its citric acid content is high and calcium content is low (9, 20, 26).

*V. opulus* is known to be widely used in Turkey, especially in stone disease. This plant is a species within the Caprifoliaceae (Honeysuckles) family within the Dipsacales order. The plant’s trunk, bark and fruits are utilized in pharmacology and as food in the form of pickles, jams, and in various other ways (27). In Central Anatolia, Turkey, especially in the city of Kayseri, it is widely termed as Gilaburu. There are no studies in literature about the mechanism of action of *V. opulus* in stone disease. We believe that it may be active by two ways: 1. Antioxidant properties 2. Possible citrate and potassium content of the plant.

Several studies have been carried in recent years establishing that *V. opulus* has a high potential of antioxidant activity and antimicrobial characteristics depending on the composition of the substances it contains (13-15, 28). It has also been shown to have beneficial effects in the gastrointestinal mucosa thanks to its antioxidant properties (29).

Comparison of citrate and potassium content of *V. opulus*, which has also antioxidant properties, to citrate and potassium content in lemon juice was the purpose of this study. We demonstrated that if *V. opulus* is citrate- and potassium-rich, therefore in relation to both its antioxidant properties and its high content in citrate and potassium, it could be argued that it can be an alternative to pharmacological agents in the treatment of hypocitraturic stone patients. In our study it has been found that potassium content in *V. opulus* content is higher than that of lemon juice whereas calcium content is lower. No statistically sig-
significant difference from lemon juice in terms of the content of citrate was observed. This result suggests us that V. opulus may have a citraturic effect as much as lemon juice and can also provide an alkali load due to its high content of potassium. This alkali load increases urinary citrate excretion by reducing renal tubular reabsorption and citrate metabolism. In addition to alkalinizing urine, alkali load also affects citrate reabsorption from the kidneys. The low calcium and sodium content of V. opulus could also be considered as an advantage for stone patients.

Conclusions
In our study, we have identified citrate and potassium V. opulus content as high as that in lemon juice. Due to its antioxidant properties as well as to its high content of both citrate and potassium, V. opulus can be recommended to stone patients. We think that it is advisable just as lemon or orange juice in mild-to-moderate hypocitruria as an alternative to potassium citrate. However, clinical trials on this subject are desirable.

References

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